#### **BIOCHEMISTRY-I**

#### **Course Objectives:**

- 1. To provide knowledge about macro molecules of eukaryotic cells and organelles, including membrane structure and dynamics;
- 2. To provide in-depth knowledge about the polymerized organic compounds of life.
- 3. To provide knowledge of the principles of bioenergetics and enzyme catalysis
- 4. To provide knowledge of the chemical nature of biological macromolecules, their threedimensional structure, and the principles of molecular recognition;

#### **Course Learning Outcome**

By the end of the course, students should be able to:

- 1. Demonstrate knowledge and understanding of the molecules of living cells;
- 2. Demonstrate knowledge and understanding of the principles that govern the structures of macromolecules and their participation in molecular recognition;
- 3. Use basic laboratory skills and apparatus to obtain reproducible data from biochemical experiments;
- 4. Implement experimental protocols, and adapt them to plan and carry out simple investigations;
- 5. Analyse, interpret, and participate in reporting to their peers on the results of their laboratory experiments;

6. Participate in and report orally on team work investigations of problem- based assignments;

## **Course Contents**

- 1. Introduction to Macromolecules
- Structure, types and role of various building blocks their respective macromolecules.
- Carbohydrates: Introduction; Classification Stereoisomerism, Structure, types and role of monosaccharides, oligosaccharides and
- polysaccharides; Glycosaminoglycans and glycoconjugates;
- Carbohydrates as an information carrier molecule.
- 2. Amino acids, peptides & proteins:
- Types of amino acids & their classification;
- Uncommon amino acids; Acid/base behavior of amino acids.
- Titration curves in amino acids and their importance:
- Peptides & proteins;
- Biologically active peptides & polypeptides;
- Amino acid sequence in proteins & their importance; Conjugated proteins;
- **3.** Purification Techniques for Proteins
- An outline of purification techniques for proteins; column chromatography, gel electrophoresis; Isoelectric focusing;
- 4. Organization of proteins:
- Structural levels of proteins; Covalent structure of proteins;
- function of some structural & functional proteins; Hemoglobin, Cytochrome-c: Chymotrypsin, alpha Keratin and Collagen;
- Proproteins, their examples and role;
- 5. Enzymes
- Enzymes, their importance, classification & nomenclature, Function & inhibition.
- 6. Lipids:

- Introduction & classification of lipids; Fatty acids, their types; Storage lipids;
- 7. Classification and important characteristics;
- Triacyclglycerols; waxes Structural/membrane lipids; Glycerophospholipids with Ether andEster linkages Galactolipids & Sulfolipds: Sphingolipids their types & importance: Sterols, their structure, types & functions. Examples of Functional diversity of Lipids as Signaling molecules, Cofactors, Electron carrier, antioxidants, pigments etc.

# 8. Nucleic acids

- Nucleic acids and their types; Structure and role of various Bases in nucleic acids,
- Nucleoside & Nucleotides;
- Structure of DNA and RNA molecules;
- Organization and Chemistry of Double helical structure of DNA with their details.

# **Practical:**

- 1. Preparation of standard curve for glucose by *ortho*-Toluidine method.
- 2. Estimation of glucose from blood serum or any other fluid using *ortho*-Toluidine technique.
- 3. Tests for detection of carbohydrates in alkaline medium.
- 4. Tests for detection of carbohydrates in acidic medium.
- 5. Tests for detection of Disaccharides.
- 6. Tests to demonstrate relative instability of glycosidic linkage in Disacchaide (Sucrose) & polysaccharide (Stanch).
- 7. Detection of Non-Reducing sugars in the presence of reducing sugars.
- 8. Demonstration of Acid Hydrolysis of Polysaccharide.
- 9. Determination of pKa values of an amino acid by preparation of titration curves.
- 10. Preparation of standard curve of proteins by Biuret method.
- 11. Estimation of any unknown concentration of protein using Biuret technique.

#### **Teaching-Learning Strategies**

Teaching will be a combination of class lectures, class discussions, and group work. Short videos/films will be shown on occasion.

#### Assignments

The sessional work will be a combination of written assignments, class quizzes, presentation, and class participation/attendance.

# Assessments and Examination

Sessional Work: 25 marks

Midterm Exam: 35 marks

Final Exam: 40 marks

## **Books Recommended:**

- 1. Lehninger principle of biochemistry by David L.Nelson and Michael M.Cox, 7<sup>th</sup> latest edition,ISBN-10:1-4641-2611-9,ISBN-13:978-14641-2611-6
- 2. Biochemistry by Jeremy M. Berg , John L. Tymoczko; Lubert Stryer ,ISBN-10:1429229365,ISBN-13:97814229229364
- 3. Berg, J. M., Tymoczko, J. L., Lubert Stryer. 2010. Biochemistry. 7th Ed.
- 4. Lodish, H., Berk, A., Zipursky, S. L., Paul. M., Baltimore D, Darnell, J. 2012. Molecular Cell Biology.
- 5. David L. Nelson, and Michael M. Cox, 2000. Lehninger Principles of Biochemistry, 3rd Ed., Macmillan Worth Publishers, New York.
- 6. Murray, R.K., Granner, D.K., Mayer, P.A. and Rodwells, V.W., 2000. Voet. D., Voet, J.G., and Pratt, C.W., 1999. Fundamentals of Biochemistry, John Wiley and Sons, Inc., New York.
- 7. Zubay, G., 1995. Biochemistry, 4th Ed., Wm. C. Brown Publishers, Inc., Oxford, England.
- 8. Stryer, L., 1995. Biochemistry, 6th Ed., W.H. Freeman and Company, New York.
- 9. Nelson, D. L., Cox, M. M. 2012. Lehninger Principles of Biochemistry. McMillan Worth Publishers, New York.
- 10. McKee, T., McKee, J.R. 2003.Biochemistry:

- 11. The Molecular Basis of Life. 3<sup>rd</sup> Edition, McGraw-Hill
- 12. Lodish, H., Berk, A., Zipursky, S. L., Paul. M., Baltimore D, Darnell, J. 2012. Molecular Cell Biology.
- 13. McKee, T., McKee, J.R. 2003.Biochemistry:
- 14. The Molecular Basis of Life. 3<sup>rd</sup> Edition, McGraw-Hill
- Molecular cell biology W.H Freeman by Lodish, Berk, Krieger, Scott, Bretscher, Ploegh and Matsudaira 8<sup>th</sup> edition/latest edition,ISBN:1464183392,ISBN-13:97814641183393

# **Text book for Practical:**

- 1. Plummer, David T., 1990. An Introduction to Practical Biochemistry, 4<sup>th</sup> Ed. McGraw-Hill Book Company, London.
- 2. Wilson, K and Walker, J., 1994. Practical Biochemistry: Principles and Techniques, 4<sup>th</sup> Ed., Cambridge University Press.
- 3. Sawhney, S.K and Singh, R., 2008. Introductory Practical Biochemistry, Narosa Publishing House, New Delhi, India.